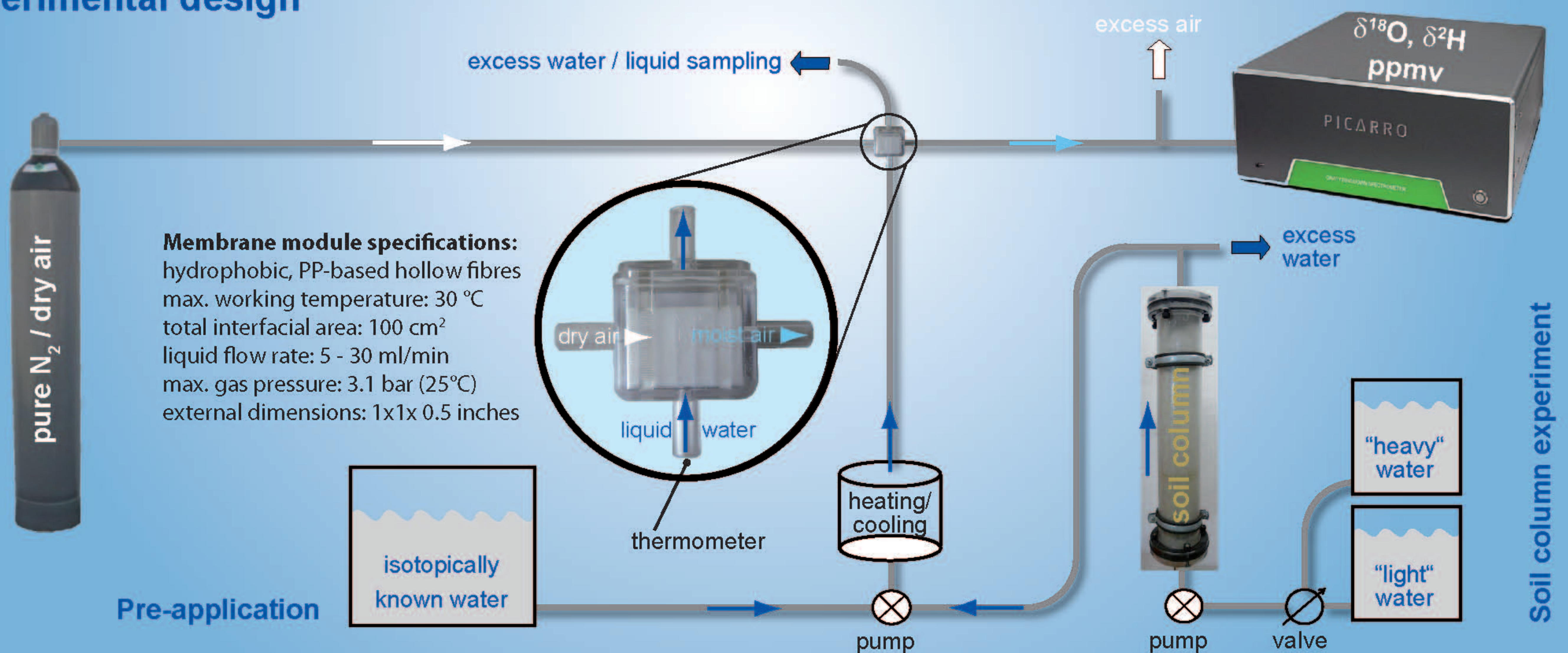


Introduction

- Isotope studies are still a trade-off between limited spatio-temporal resolution and extensive lab work
- In conventional isotope analytics a significant time lag exists between sampling and data acquisition (unlike EC or T measurements)
- Laser-based analyzers are now available and capable of measuring stable water isotopes in the vapor phase directly and continuously

► **Challenge: Convert liquid water to water vapor and continuously provide it to analyzer**

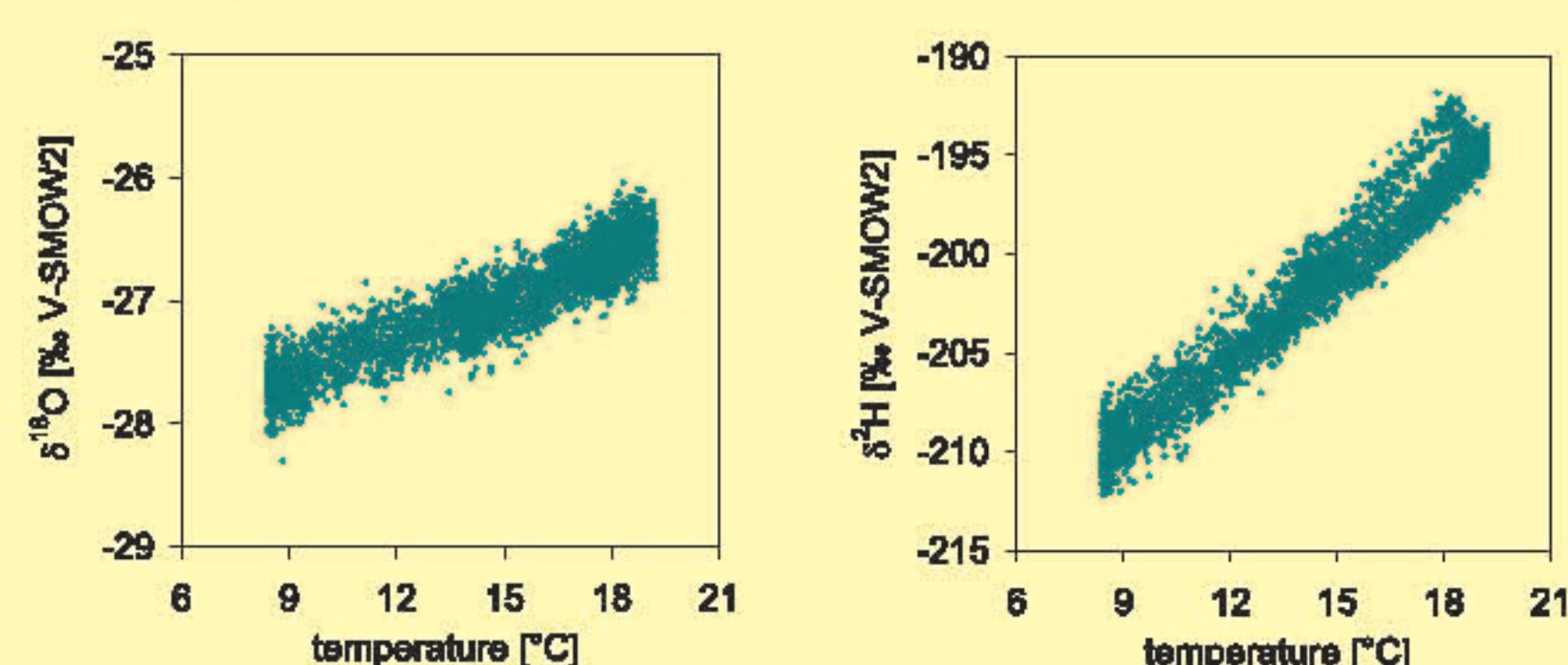
Experimental design



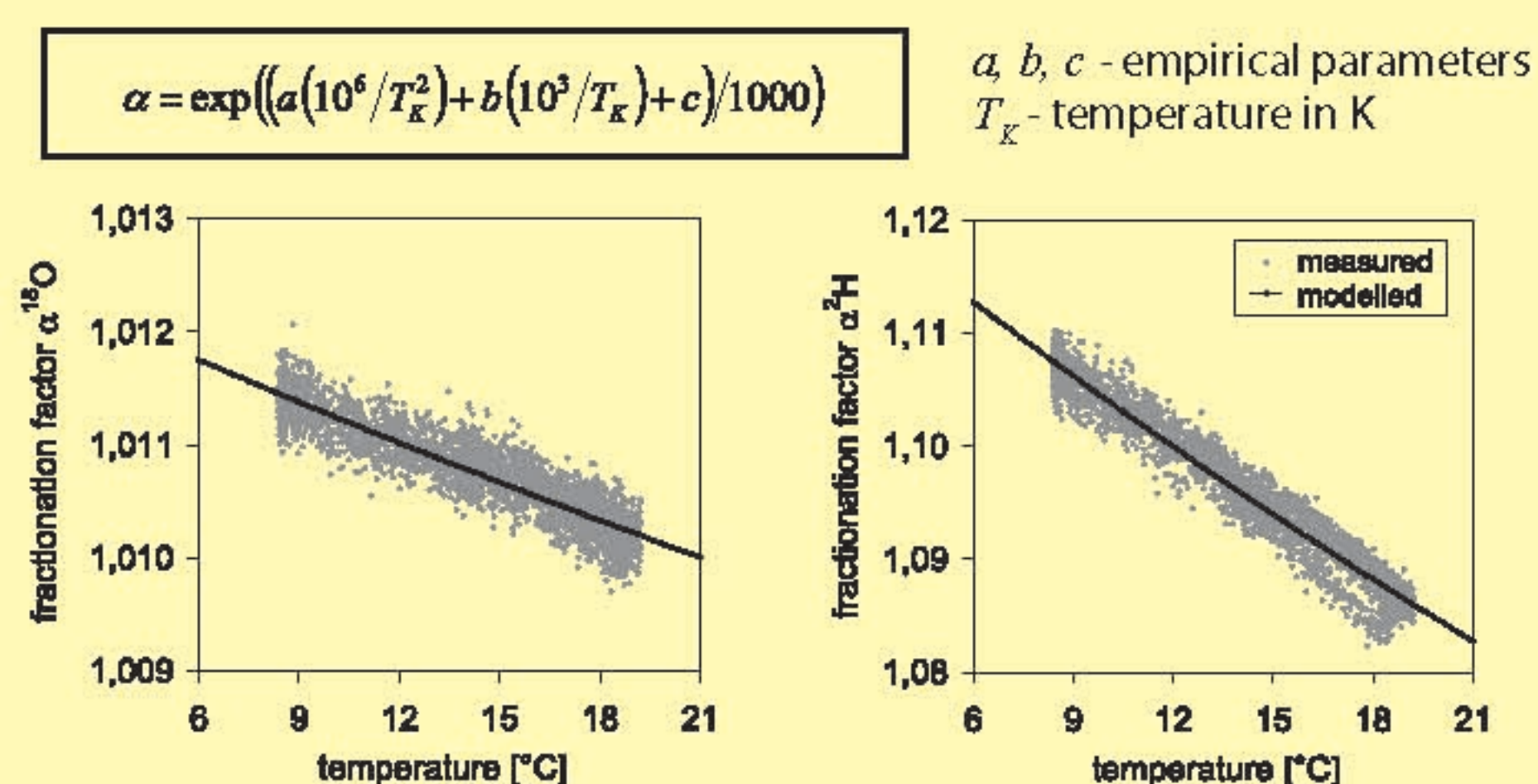
Pre-application steps

Determination of membrane-specific fractionation factors

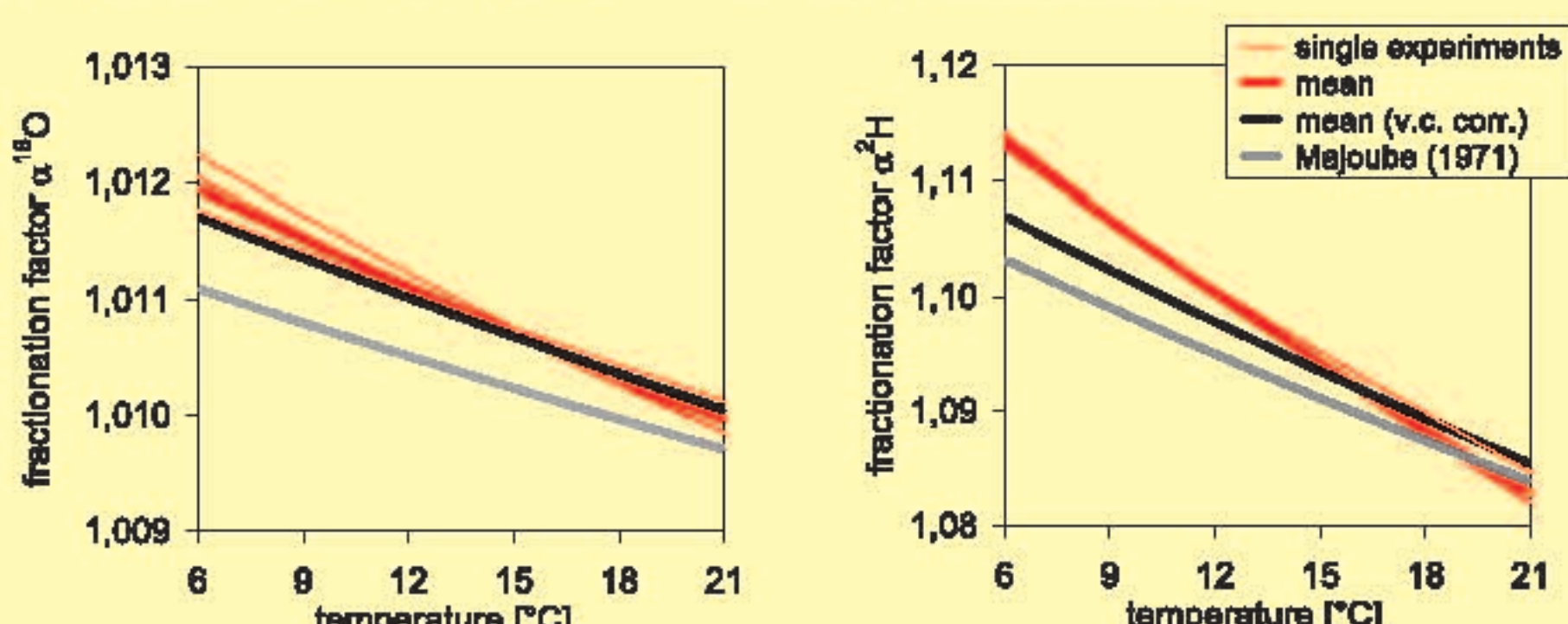
1. Measuring of thermal dependency of water vapor isotopes and calibration



2. Calculation and modelling of fractionation factors α

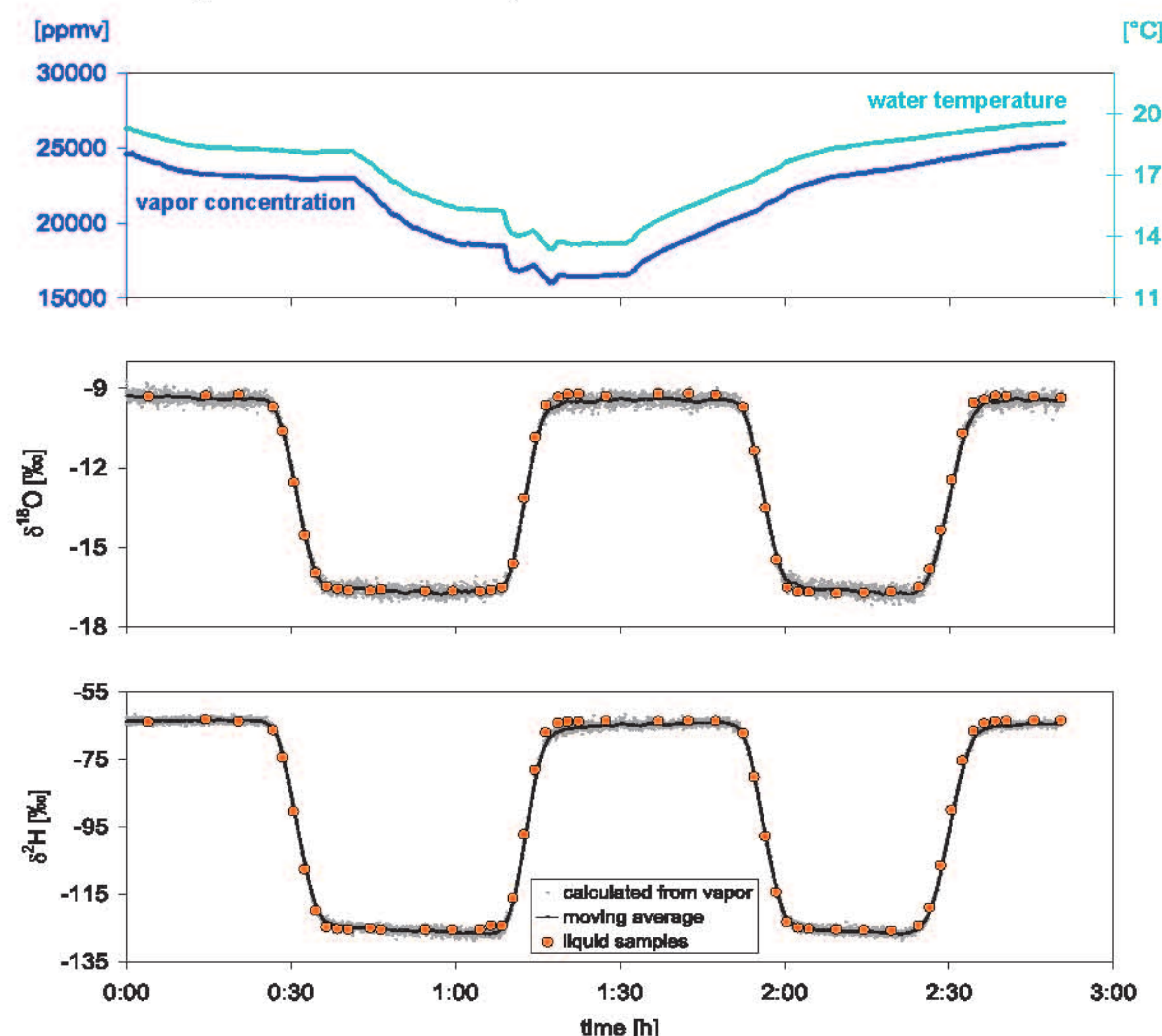


3. Repetition and comparison with free water surface



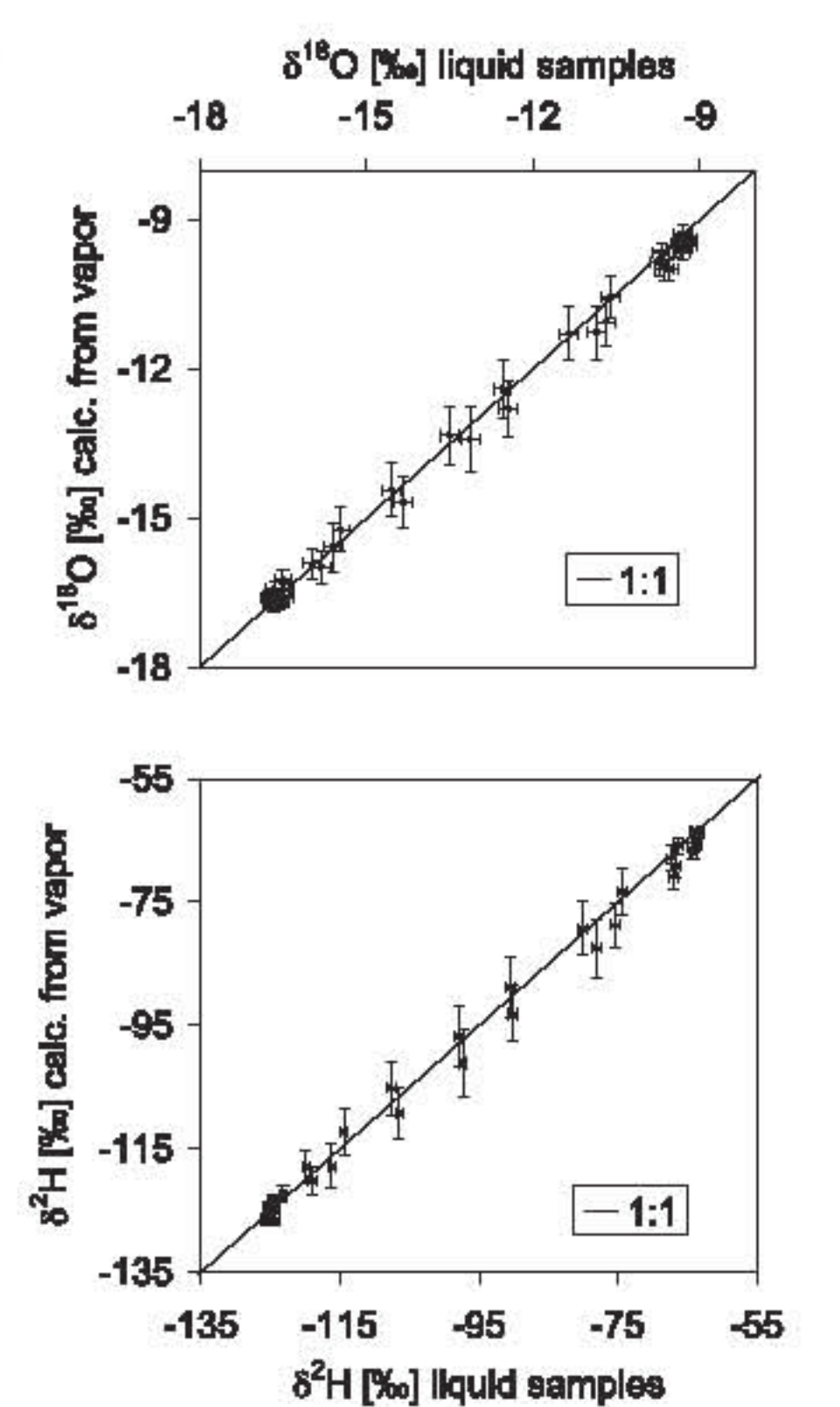
Application: soil column experiment

Testing the method under highly unstable conditions (isotopically, thermally)
 Verification by conventional analysis



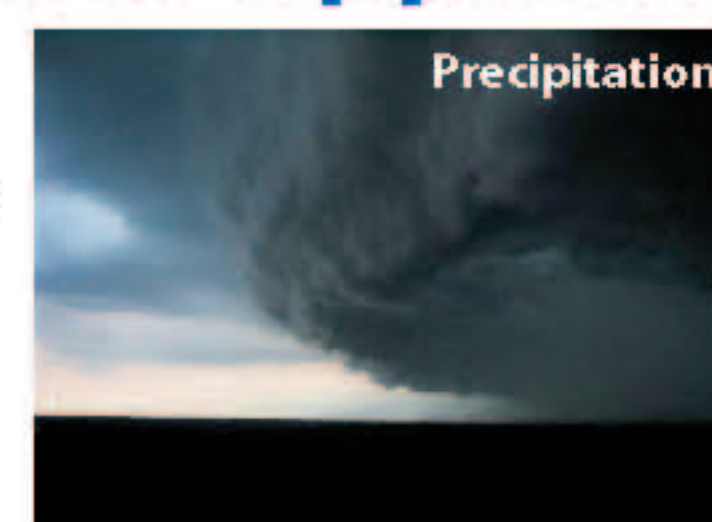
Soil column parameters:
 length: 50 cm
 diameter: 10 cm
 filling: fine sand
 water flow rate: 27 ml/min

Flow rates at membrane:
 water: 22ml/min
 gas (N₂): 195 ml/min



Possible further applications

Suitable wherever dynamic processes have to be observed in real time and with high temporal resolution



Conclusion

- Hydrophobic membranes may have specific isotopic fractionation factors
- The proposed method provides real-time data and captures even abrupt changes

Response time: about 10s (depending on setup dimensions and flow rates)

Resolution: minutes or below

Precision: comparable to conventional analysis (0.16‰ for $\delta^{18}\text{O}$, 1.1‰ for $\delta^2\text{H}$)

Supervision: minimum requirements

Restrictions: water temperature must not exceed ambient temperature (else: heating / dilution)

